

```

p2=25000
T2=600
h2=enthalpy(STEAM_NBS,p=p2,T=T2)
s2=entropy(STEAM_NBS,p=p2,T=T2)
h3i=enthalpy(STEAM_NBS,p=p3,s=s2)
{p3=5000}
p4=p3
eta_t=0.84
h2-h3=(h2-h3i)*eta_t
T3=temperature(STEAM_NBS,h=h3,p=p3)
h4=enthalpy(STEAM_NBS,p=p4,T=T2)
s4=entropy(STEAM_NBS,p=p4,T=T2)
h5i=enthalpy(STEAM_NBS,p=p5,s=s4)
p5=5
h4-h5=(h4-h5i)*eta_t
x5=quality(STEAM_NBS,p=p5,h=h5)
p6=p5
h6=enthalpy(STEAM_NBS,p=p6,x=0)
s6=entropy(STEAM_NBS,p=p6,x=0)
p1=p2
h1i=enthalpy(STEAM_NBS,p=p1,s=s6)
eta_p=0.70
h1-h6=(h1i-h6)/eta_p
wp=h6-h1
wt_hp=h2-h3
wt_lp=h4-h5
qin=(h2-h1)+(h4-h3)
eta_th=(wt_hp+wt_lp+wp)/qin
HR=1/eta_th*K1
K1=3600

```

## SOLUTION

Unit Settings: [kJ]/[C]/[kPa]/[kg]/[degrees]

Maximization of eta\_th(p3) 17 iterations: Quadratic Approximations method

$\eta_p = 0.7$	$\eta_t = 0.84$	$\eta_{th} = 0.4078$ [-]	$h1 = 173.4$ [kJ/kg]	$h1i = 162.7$ [kJ/kg]
$h2 = 3490$ [kJ/kg]	$h3 = 3014$ [kJ/kg]	$h3i = 2923$ [kJ/kg]	$h4 = 3679$ [kJ/kg]	$h5 = 2496$ [kJ/kg]
$h5i = 2271$ [kJ/kg]	$h6 = 137.7$ [kJ/kg]	$HR = 8827$ [kj/kwh]	$K1 = 3600$ [kj/kwh]	$p1 = 25000$ [kPa]
$p2 = 25000$ [kPa]	$p3 = 3416$ [kPa]	$p4 = 3416$ [kPa]	$p5 = 5$ [kPa]	$p6 = 5$ [kPa]
$qin = 3982$ [kJ/kg]	$s2 = 6.359$ [kj/kg-k]	$s4 = 7.446$ [kJ/kg-K]	$s6 = 0.4761$ [kj/kg-k]	$T2 = 600$ [C]
$T3 = 313.2$ [C]	$wp = -35.71$ [kJ/kg]	$wt_{np} = 476.7$ [kJ/kg]	$wt_p = 1183$ [kJ/kg]	$x5 = 0.9733$ [-]

No unit problems were detected.

Purple units were automatically set. Right click on the variable to confirm or change the units.

